1) The word "RADICAL" comes from the Latin word for "root". In the olden days "RADICAL" referred to a fragment of a molecule that remained unchanged throughout a series of reactions (e.g. the use of R-OH to represent a variety of alcohols). The term "FREE RADICAL" then was later introduced to refer to a molecular fragment that was NOT bonded to anything else. Today, the terms radical and free radical are used interchangeably.

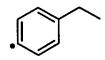
What is our modern scientific definition of a (free) radical species?

- 2-5) Circle the *most stable* species in each threesome.
- 2) CH₄
- ĊН<sub>3</sub>
- ⊕ CH<sub>3</sub>

- 3)
- **/**
- **/**

- 4)
- $\dot{\bigcirc}$

5)







- 6-7) Bond Dissociation Energy (BDE) is the energy required to break a covalent bond homolytically.
- What is a covalent bond?
- What does homolytically mean?

8) Draw in all 14 Hydrogen atoms on this line angle diagram.

9-13) For each molecule below, indicate which C-H bond is the *weakest* (meaning *lowest* Bond Dissociation Energy).

14) Briefly explain the following (correct) observation – "steric hindrance can be used to produce especially stable (relatively long lived or persistent) radical species.

15-18) Label each reaction below as an Addition; Abstraction; Dimerization; Disproportionation or Fragmentation.

18) 
$$H_3C \xrightarrow{O} O^{\bullet}$$
  $\longrightarrow$   $\bullet CH_3 CO_2$ 

19) What must occur in the initiation step of a free radical reaction?

20) What is meant by the term "free radical inhibitor"?

21-25) Use appropriate curly (FISH HOOK) arrows to describe the electron movement for the following processes.

21) 
$$R^{O_{O}}R$$
  $Heat$  2  $R-O^{\circ}$ 

22-23)  $Heat$  2  $R-O^{\circ}$ 

22-23)  $Heat$  2  $R-O^{\circ}$ 

24)  $Heat$  2  $R-O^{\circ}$ 
 $R^{O_{O}}R$   $R^{O_$ 

BONUS POINT: As stated in Q1, the term "RADICAL" comes from the Latin word for "root". What edible root vegetable gets its name from this word? (Hint: it is typically a crunchy salad vegetable).

1) The word "RADICAL" comes from the Latin word for "root". In the olden days "RADICAL" referred to a fragment of a molecule that remained unchanged throughout a series of reactions (e.g. the use of R-OH to represent a variety of alcohols).

The term "FREE RADICAL" then was later introduced to refer to a molecular fragment that was NOT bonded to anything else. Today, the terms radical and free radical are used interchangeably.

What is our modern scientific definition of a (free) radical species?

A species with one unpaired valence electron.

2-5) Circle the *most stable* species in each threesome.

2)

ĊН<sub>3</sub>

3)



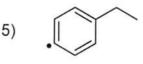
4)

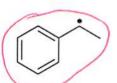






5)







6-7) Bond Dissociation Energy (BDE) is the energy required to break a covalent bond homolytically.

- What is a *covalent bond*?



- What does *homolytically* mean?

8) Draw in all 14 Hydrogen atoms on this line angle diagram.

9-13) For each molecule below, indicate which C-H bond is the *weakest* (meaning *lowest* Bond Dissociation Energy).

14) Briefly explain the following (correct) observation – "steric hindrance can be used to produce especially stable (relatively long lived or persistent) radical species.

Steric hindrence will slow down (or stop) the reactions of a free radical with other species through electron pair-electron pair (epulsion.

15-18) Label each reaction below as an *Addition*; *Abstraction*; *Dimerization*; *Disproportionation* or *Fragmentation*.

15) O=0

Abstraction

H-Br

17) Pho I<sub>2</sub>

Ph-I Io

18) H<sub>3</sub>C Oo

19) What must occur in the *initiation* step of a free radical reaction?

The generation of radicals

20) What is meant by the term "free radical inhibitor"?

Something that will reach with reacher radicals to produce a much more stable (or unreachers) species.

21-25) Use appropriate curly (FISH HOOK) arrows to describe the electron movement for the following processes.

21) 
$$R \stackrel{\text{Heat}}{\circ} 2 R - 0 \stackrel{\bullet}{\circ}$$

22-23)  $\frac{\text{Heat}}{\text{CN}} 2 R - 0 \stackrel{\bullet}{\circ}$ 

22-23)  $\frac{\text{Heat}}{\text{CN}} 2 \stackrel{\bullet}{\text{CN}} N \equiv N$ 

24)  $\frac{\text{Heat}}{\text{CN}} 2 \stackrel{\bullet}{\text{CN}} N \equiv N$ 

25)  $\text{Bu}_3 \text{Sn} \stackrel{\bullet}{\bullet} \text{Br} \longrightarrow \text{Bu}_3 \text{Sn-Br}$ 

BONUS POINT: As stated in Q1, the term "RADICAL" comes from the Latin word for "root". What edible root vegetable gets its name from this word? (Hint: it is typically a crunchy salad vegetable).

RADISH