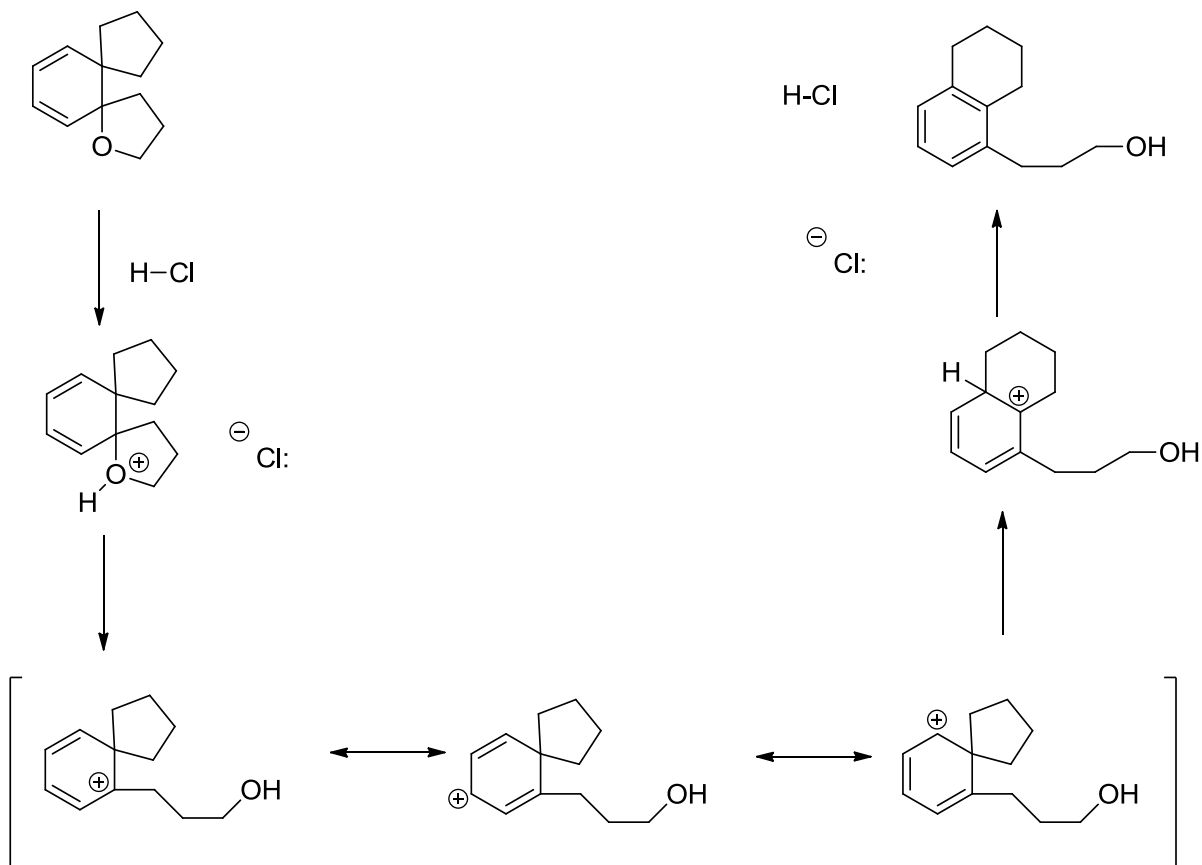
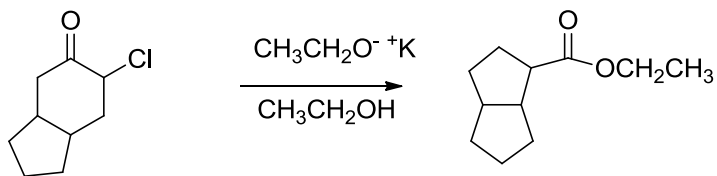


1) Draw in the curly arrows for this reaction. (7pts)

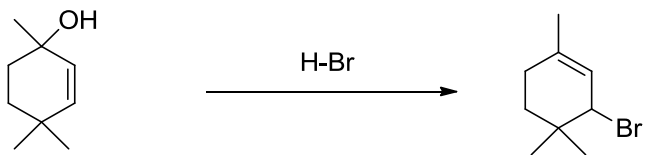


2) Draw a reasonable mechanism for this process. (7pts)

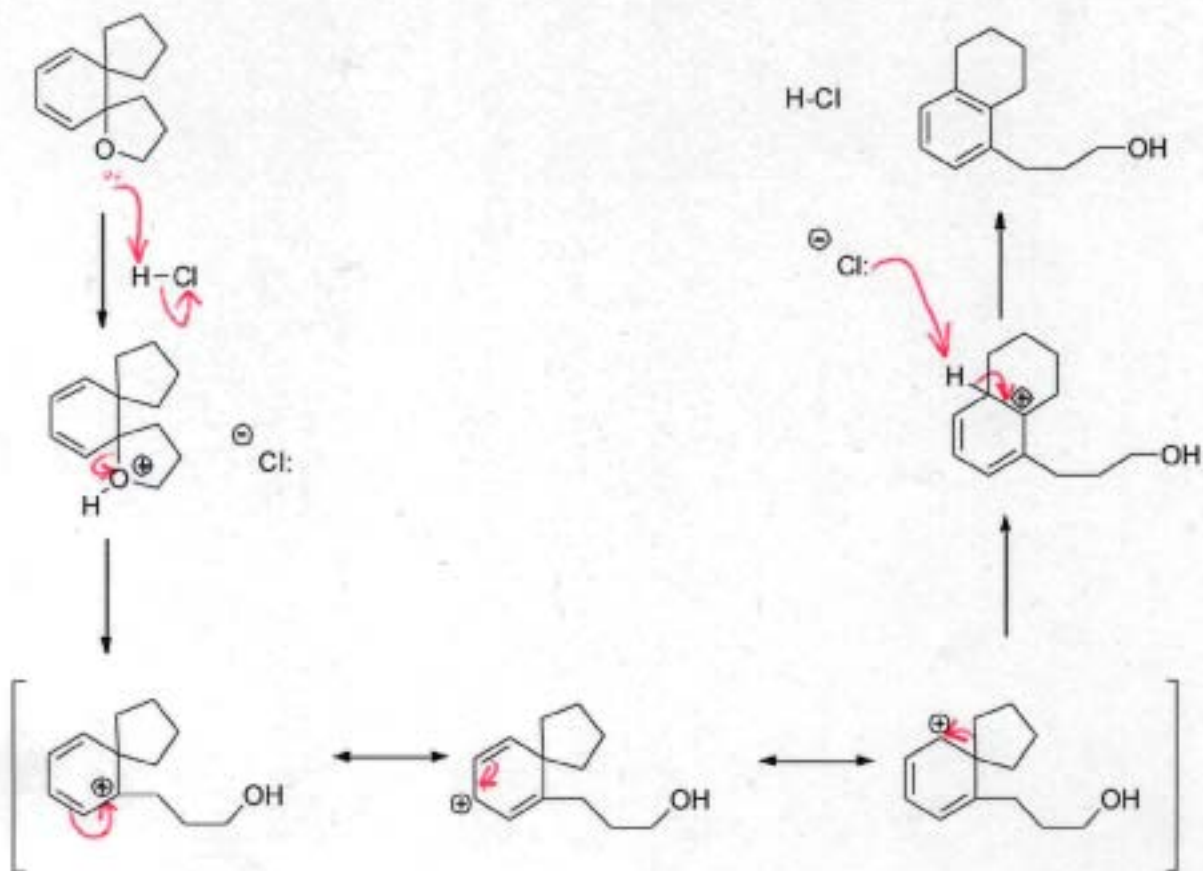


Bonus point if you can draw the product that would be produced if this reaction was performed using KOH in DMSO instead of ethoxide/ethanol. (Another bonus point if you get the correct ionic form of the product).

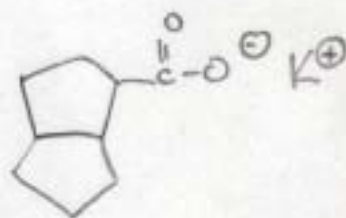
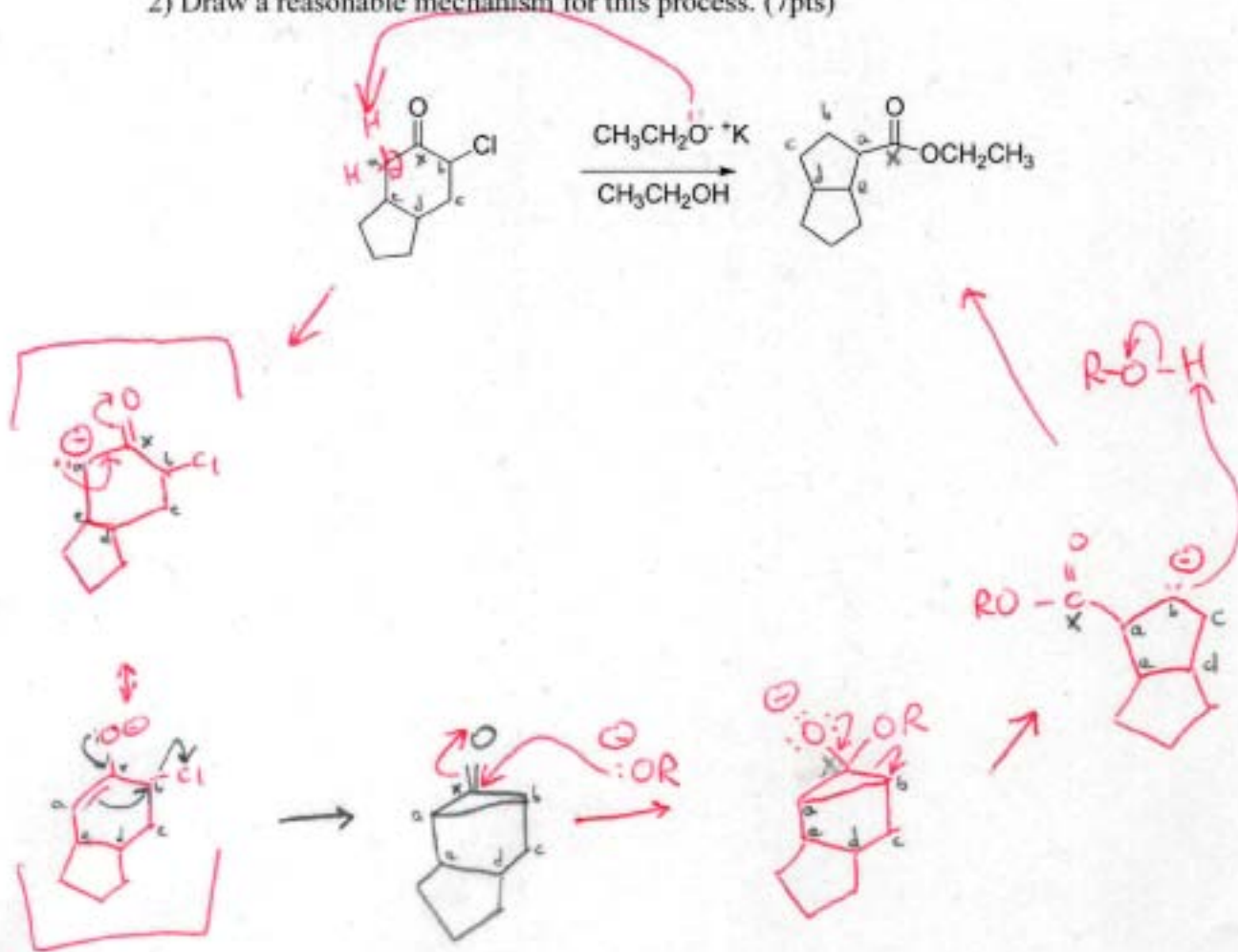
3) Write the mechanism for the following reaction. (6pts)



1) Draw in the curly arrows for this reaction. (7pts)



2) Draw a reasonable mechanism for this process. (7pts)



carboxylic acid
is deprotonated
in basic
conditions.

Bonus point if you can draw the product that would be produced if this reaction was performed using KOH in DMSO instead of ethoxide/ethanol. (Another bonus point if you get the correct ionic form of the product).

3) Write the mechanism for the following reaction. (6pts)

